

# Some Properties of TiO<sub>2</sub>/Au Nanocomposite Thin Films Produced by Spincoating Method for Application in Plasmonic Solar Cells

Nguyen Duy Thien<sup>1,\*</sup>, Nguyen Tien Thanh<sup>1</sup>,  
Nguyen Thi Thuy<sup>2</sup>, Le Van Vu<sup>1</sup>, Dao Khac An<sup>2</sup>

<sup>1</sup>*Centre for Materials Science, Faculty of Physics, VNU University of Science,  
334 Nguyen Trai, Thanh Xuan, Hanoi, Vietnam*

<sup>2</sup>*Energy Materials and Devices Lab, Institute of Materials Science, (VAST),  
18 Hoang Quoc Viet, Cau Giay, Hanoi, Vietnam*

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**Abstract:** The TiO<sub>2</sub>/Au nanocomposite thin films play a very important role in the plasmonic solar cell. In such as composite systems, the localized surface plasmon resonance (LSPR) of embedded metal nanoparticles determines the light absorption enhancement of the semiconductive part (TiO<sub>2</sub>); hence, enhances the efficiency of the plasmonic solar cell. The characterizations of the TiO<sub>2</sub>/Au nanocomposite thin films – such as light absorption, scattering and LSPR capacities - depend on many parameters: sizes, weight ratio of metal nanoparticles, thickness of the thin film layers and the arrangements of TiO<sub>2</sub>/Au composite in integrated-matrix system. In this work, we outline the preparation method of the TiO<sub>2</sub>/Au nanocomposite thin films, then we investigate the effect of technological conditions – such as TiO<sub>2</sub>/Au mixture solution concentration, TiO<sub>2</sub>/Au weight ratio and number of spin-coating layers – to the surface morphology of the nanocomposite to find out the optimum fabrication conditions. Some experimental results of thin film such as structural property (X ray diffraction, thin film morphology...) and optical property (absorption coefficient concerning LSPR...) will also be shown and discussed.

*Keywords:* TiO<sub>2</sub>/Au thin film, LSPR, plasmonic solar cell.

## 1. Introduction

Recently, fossil energy sources are depleted due to the increasing exploitation of the countries. Beside, the extraction and over usage have caused many consequences such as environmental pollution, the greenhouse effect and climate change. Therefore, one of serious challenges to mankind is to ensure the energy security or energy sustainability. This challenge has to be answered with a

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\* Corresponding author. Tel.: 84- 915451155  
Email: duythien\_303@yahoo.com

solution to produce the renewable energy sources, One of them is the solution transforms of solar energy into electrical energy based on photovoltaic cells [1-4]. Up to now there are three solar cell generations have been developed, among them the plasmonic solar cells, belonging to the third generation, are promising and that is promise new type and they are getting the most attention of scientists [3-8, 9-10]. In the plasmonic solar cells, the most important structure is the structure of the integrated noble metal nanoparticle ( $\text{TiO}_2/\text{Au}$  (Ag...)) system. This, so called plasmonic nanostructures where the plasmonic resonance induces a dipole on the metal nanoparticle, enhance electric field around the metal nanoparticle, could support the formation of the surface plasmon resonance in response to a photon flux, localizing electromagnetic energy close to their surfaces. The enhanced photovoltaic properties from the incorporation of metal nanoparticles/ semiconductor have been studied for various systems [5 -8,11,12]. So far, the plasmonic solar cells are developed at the beginning stage that many research issues, including both theoretical and practical problems must be overcome [5-8, 11-13].

In previous reports, we have also designed and produced some samples of the plasmonic solar cells, investigated the structural, electrical and optical properties, however, the obtained results of electrical and optical properties were still limited [14]. In this work we focused on the spincoating method for fabricating  $\text{TiO}_2/\text{Au}$  thin film through two steps: preparation of  $\text{TiO}_2/\text{Au}$  sol solution and using this sol solution for spincoating to create  $\text{TiO}_2/\text{Au}$  thin films, after that we have investigated the effect of technological conditions such as  $\text{TiO}_2/\text{Au}$  solution concentration,  $\text{Au}/\text{TiO}_2$  weight ratio and number of spin-coating layers on structural and optical properties of the  $\text{TiO}_2/\text{Au}$  thin films.

## 2. Experiment

All the precursors with high purity were purchased from commercialized distributors: Chloroauric acid 99.98 % ( $\text{HAuCl}_4$ ), Sodium borohydrid 99.8% ( $\text{NaBH}_4$ ) from MERCK, German; pure Titanium dioxide nanoparticles ( $\text{TiO}_2$ ) from Sigma, America and polyvinyl pyrrolidone 98% (PVP – MW 30000 Dal) from Korea. The  $\text{TiO}_2/\text{Au}$  composite thin films were prepared by sol-thermal calcinations on  $1\text{cm} \times 2\text{cm}$  ITO substrates (from France) and sheet resistance of 200-250  $\Omega/\text{square}$ .

Firstly, to create 5% of  $\text{Au}/\text{TiO}_2$  weight ratio composite (labeled as  $\text{TiO}_2/\text{Au}$  5%): 25 mg of  $\text{TiO}_2$  was dispersed into 25 ml of  $\text{HAuCl}_4$  1mM, which contained 250 mg PVP. The mixture was gently mixed by magnetic stirring to obtain a homogenous aqueous solution. After that, we added a small amount of ice-cold 0.01M  $\text{NaBH}_4$  to  $\text{Au}^{3+}$  from chloroauric acid to  $\text{Au}^0$ . The solution color changed to a typical dark red. We have used centrifugation to get higher concentration solution, then the solution was dispensed in ethanol – later called xerogel, which contained the mixture of  $\text{TiO}_2$  and Au nanoparticles. Secondly, we created  $\text{TiO}_2/\text{Au}$  thin film xerogel on ITO substrate by spin-coating method. In particular, 50  $\mu\text{L}$  of xerogel solution was deposited on ITO plate spinning at 3000 rpm for 30s. In the next step- calcinations step, the remained organic precursors and unnecessary reaction products were evaporated substrate by heating process at  $450^\circ\text{C}$  for 1hour. Finally we received  $\text{TiO}_2/\text{Au}$  nanocomposite thin films. Thin films 10% of  $\text{Au}/\text{TiO}_2$  weight ratio composite thin film

(labeled as TiO<sub>2</sub>/Au 10%) was prepared by the same method, where the used amount of HAuCl<sub>4</sub> 1mM solution was 50mL.

In this paper We selected the TiO<sub>2</sub>/Au 5% composite for next investigations. First, different concentrations TiO<sub>2</sub>/Au 5% composite were prepared by increasing the amount of added TiO<sub>2</sub> nanoparticles, while the weight ratio of Au/TiO<sub>2</sub> was kept constant. Typically, the mixture of TiO<sub>2</sub>/Au in PVP in ethanol was diluted to different concentrations (Table 1). The concentrations of the samples were defined by the precursor TiO<sub>2</sub> colloids in xerogel solution.

Table 1. The precursor amounts used in different the composite thin films on ITO substrates (samples) for investigating the effect of the TiO<sub>2</sub>/Au precursor concentration on the filling efficiency

Sample	Amount of precursor TiO <sub>2</sub> nanoparticles in xerogel solution	Amount of initial TiO <sub>2</sub> /Au mixture before centrifugation	Amount of added ethanol in xerogel
M1	10mg/ml	25ml	2500 μl
M2	40mg/ml	25ml	625 μl
M3	50mg/ml	25ml	500 μl

The crystal structure of the synthesized products was analyzed by X-ray diffraction (XRD) using X-ray diffractometer Siemens D5005, Bruker, Germany, with Cu-K<sub>α1</sub> radiation ( $\lambda = 0.154056$  nm). The surface morphology of the samples was investigated by using a Nova NANOSEM 450 scanning electron microscope and Dektak 150 Profiler, Veeco, USA. The UV-vis absorption spectra were measured by Shimadzu UV 2450 PC spectrometer, Japan.

### 3. Results and discussions

#### 3.1. Crystal structure and morphology

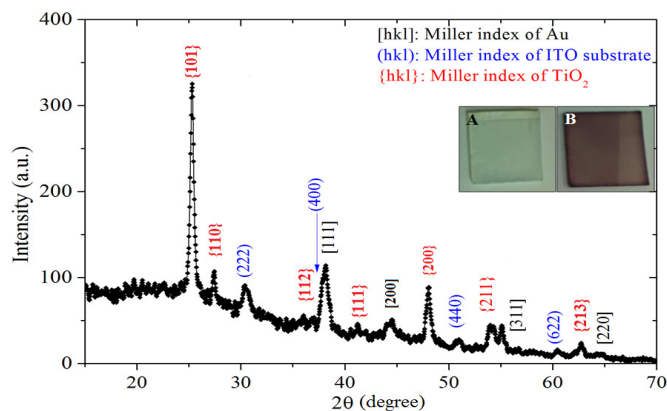


Fig. 1. X-ray diffraction pattern of TiO<sub>2</sub>/Au nanocomposite thin film prepared by calcinated spin-coating method. The weight ratio of Au:TiO<sub>2</sub> was 5%. Inset: ITO substrate without (A) and with TiO<sub>2</sub>/Au composite (B)

Fig. 1 shows XRD pattern of TiO<sub>2</sub>/Au 5% nanocomposite thin film spincoated on ITO substrates, where the inset shows the picture of the TiO<sub>2</sub> thin film (A) and TiO<sub>2</sub>/Au composite thin film (B) on ITO substrate. TiO<sub>2</sub>/Au nanocomposite thin film exhibits characteristic dark red of gold nanoparticles. The typical peaks of ITO substrate arise at 30.5°, 37.1°, 51.0° and 60.4°, corresponding to (222), (400), (440) and (622) planes, respectively. Existence of TiO<sub>2</sub> in nanocomposite shows in the peaks at 25.3°, 27.5°, 36.0°, 41.3°, 48.1°, 54.1° and 62.8°, coinciding with previous published results, which relate to {101}, {110}, {112}, {111}, {200}, {211} and (213) of anatase structure of TiO<sub>2</sub> [15]; while the presence of Au nano-colloids is demonstrated via typical peaks face-centered cubic structure of Au at 38.2°, 44.4° and 64.6° corresponding to [111], [200] and [220], respectively. (CAS: 7440-57-5) [16]

### 3.2. The effect of the TiO<sub>2</sub>/Au mixture solution concentration

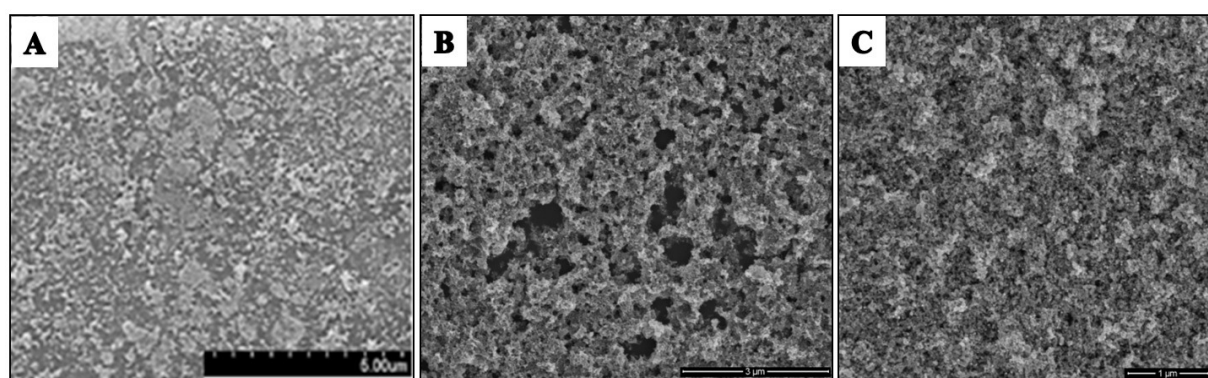


Fig. 2. SEM images of TiO<sub>2</sub> /Au nanocomposite thin films spincoated on ITO substrates with different concentrations of TiO<sub>2</sub>/Au sol solution: a) 10mg/ml, b) 40mg/ml and c) 50mg/ml.

Scanning electron microscopic (SEM) images of TiO<sub>2</sub>/Au nanocomposite thin films with different concentrations of TiO<sub>2</sub>/Au sol solution on ITO substrate are presented in the Figure 2. In M1 sample, when the solution concentration of TiO<sub>2</sub>/Au is 10mg/ml, there are a lot of voids on ITO surface (A). Total filled-up area has decreased significantly when the TiO<sub>2</sub>/Au sol solution concentrations are increased (Fig 2b). Especially, the unfilled areas disappear when TiO<sub>2</sub>/Au solution concentration are 50mg/ml (Fig 2c). In further investigations, we use this concentration of TiO<sub>2</sub>/Au precursors to synthesize TiO<sub>2</sub>/Au nanocomposite thin films.

### 3.3. Thickness control

The thickness of TiO<sub>2</sub>/Au nanocomposite thin film depends on various parameters, such as viscosity and concentration xerogel solution, rotation speed while spin-coating... In this works, we fixed all of these parameters and control thickness of films via controlling times of repeating spincoating TiO<sub>2</sub>/Au solution onto substrate.

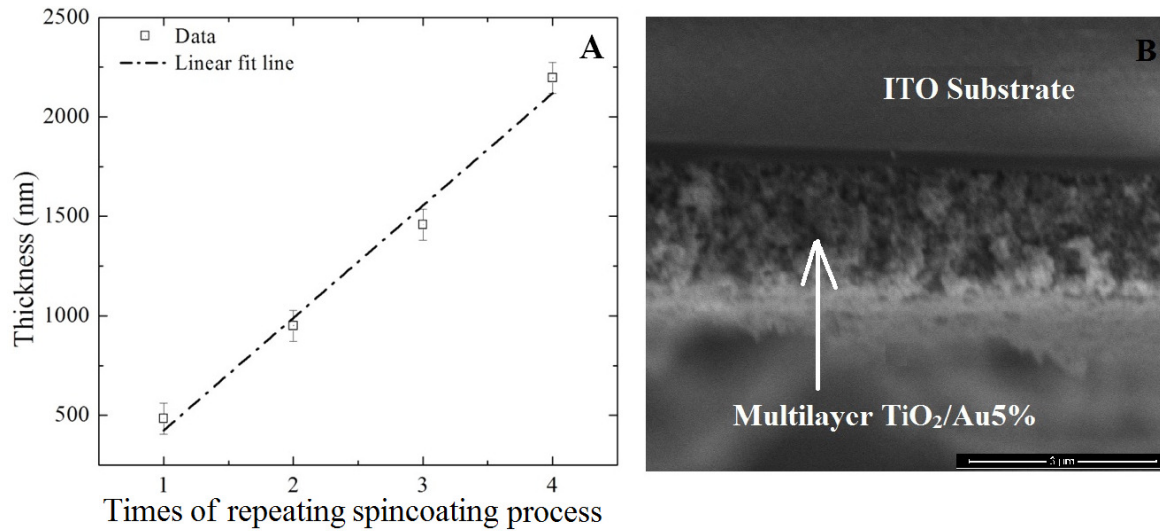


Fig. 3. The Plot dependence of the  $\text{TiO}_2/\text{Au}$  nanocomposite thin film thickness on times of repeating spincoating (a); and SEM image of four layers  $\text{TiO}_2/\text{Au}$  nanocomposite thin film (b).

As we increased the thickness of the  $\text{TiO}_2/\text{Au}$  by repeating the spin coating – calcinations process multiple times, the composite thin film could be considered as a multilayer of films and thickness of each layer was about  $480 \text{ nm} \pm 77.5 \text{ nm}$  (Fig. 3A). Table 2 shows  $\text{TiO}_2/\text{Au}$  nanocomposite thin films thickness measured by Dektak 150 equipment system when the coating time was 1, 2, 3 and 4, respectively

Table 2. Thickness of  $\text{TiO}_2/\text{Au}$  nanocomposite tin film when spin- on 1, 2, 3 and 4 layer

Multilayer	1-time coating	2-time coating	3- time coating	4-time coating
Thickness (nm)	483.0	950.0	1460.0	2190.0
Standard error (nm)	77.5	77.5	77.5	77.5

Cross-section SEM image of the 4-time spin-coated  $\text{TiO}_2/\text{Au}$  5% composite on ITO substrate is shown in Figure 3B. No layer boundary could be observed disjunction occurs. Continuous one layer thin film was formed this result implied that the obtained films were homogenous

### 3.4. Optical properties

As we know that light scattering by metal nanoparticles is of special research interest because of many applications in the energy field where nanocomposite particles are used to increase the efficiency of solar cells, the way these nanoparticles scatter light is of vital importance to the research.

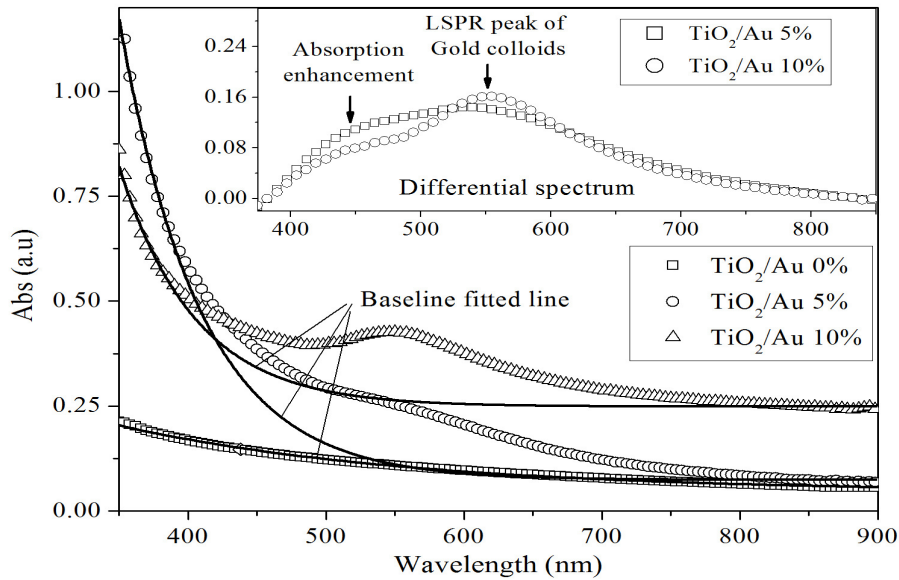


Fig. 4. The absorption spectra concerning the Localized Surface Plasmon Resonance (LSPR) of TiO<sub>2</sub>/Au 0%, TiO<sub>2</sub>/Au 5%, TiO<sub>2</sub>/Au 10% nanocomposite thin films.

Light scattering is, in general, the attenuation of a beam of light by metal nanoparticles, either by absorption or scattering. The sum of these two parts is known as the extinction. For particles with diameters well below the wavelength of light, a point dipole model describes well the absorption and scattering of light. The scattering and absorption cross-sections,  $C_{scat}$  and  $C_{abs}$ , respectively are given by the followings [17-19]:

$$c_{scat} = \frac{1}{6\pi} \left(\frac{2\pi}{\lambda}\right)^4 |\alpha|^2, \quad c_{abs} = \frac{2\pi}{\lambda} \text{Im}|\alpha|^2, \quad \text{where } \alpha = 3V \left[ \frac{\epsilon_p / \epsilon_m - 1}{\epsilon_p / \epsilon_m + 2} \right] \quad (4.1 \text{ a,b,c})$$

Here  $V$  is the particle volume,  $\epsilon_p$  is the dielectric function of the particle and  $\epsilon_m$  is the dielectric function of the embedding medium. We can see that when  $\epsilon_p = -2\epsilon_m$ , the particle polarizability will become very large. This is known as the surface plasmon resonance. This sometime is called the dipole plasmon resonance of the particle. At surface plasmon resonance condition the scattering cross-section can well exceed the geometrical cross section of the particle.

Concerning our obtained results, in general, the LSPR absorption properties of the TiO<sub>2</sub>/Au 0%, TiO<sub>2</sub>/Au 5% and TiO<sub>2</sub>/Au 10% are illustrated in Figure 4. The results show that the absorption spectra enhanced in the range of 500 nm - 600 nm and absorption intensity increases when Au concentration being in the sample increases. We used absorption of TiO<sub>2</sub> thin film without Au nanoparticle (TiO<sub>2</sub>/Au 0%) as the base level. In absorption range from 350 nm to 900nm an experimental exponential (function not shown) occurs. TiO<sub>2</sub> particles usually have good light absorption at near UV band (data not shown), then decay as the wave-length shifts to red region, where we can consider as an exponential decay. We use that decay as baseline level for data manipulation of two other spectra of

TiO<sub>2</sub>/Au 5% and of TiO<sub>2</sub>/Au 10%. An exponential decay was employed as baseline (see Fig. 4 – baseline fitted line) for the region of [350nm-400nm] and [750nm-900nm], where the typical surface Plasmon resonance of gold nanoparticles does not occur [16, 20]. Then the baselines are distracted from the original spectra to get the different spectra (Fig. 4 – inset) of TiO<sub>2</sub>/Au 5% composite and of TiO<sub>2</sub>/Au 10% composite thin films, respectively. We can see that there are enhancements arising at characteristic SPR peak of Gold nanoparticles around 540 nm. We also see there are abnormal peaks arising at near-UV region from 430 nm to 450 nm. The origin of these peaks is still unrevealed. They might relate to the phenomenon of the presence of a thin Au coat around the TiO<sub>2</sub> particles during synthesis process. AuCl<sup>4-</sup> ions from HAuCl<sub>4</sub> adsorbed on surface of the semiconductor nanoparticles during mixing and these ions were reduced to Au atom then attached on the surface of the TiO<sub>2</sub> particles meanwhile the free ions were reduced to Au nanoparticles. The presence of this enhancement also occurred at the absorption spectra of xerogel solutions (data not shown). However, more investigation are needed to clarify the origin of this enhancement.

#### 4. Conclusions

In conclusion, we have synthesized TiO<sub>2</sub>/Au nanocomposite thin film from TiO<sub>2</sub>/Au solution with different Au/TiO<sub>2</sub> weight ratio on the ITO substrates by spin coating method. Our results show that the whole ITO substrate is filled with relatively homogenous TiO<sub>2</sub>/Au composite at 50 mg/ml concentration of precursor TiO<sub>2</sub> nanoparticles in xerogel solution. Thickness of the thin film can be increased by multiple-time coating, while the homogeneity of the composite layer does not change. In our spinning conditions, each coating time would create reproducibly 480 nm ± 77 nm thick layer, which shows that we have to coat the ITO substrate at least 4 times to reach usual thickness of plasmonic solar cells.

Based on the produced TiO<sub>2</sub>/Au thin film we have investigated the absorption efficiency concerning the LSPR of the thin film. The result show that 5% of Au/TiO<sub>2</sub> weight ratio is the critical ratio to allow LSPR of the Au particles in the composite thin film. Higher amount of Au in the complex results in higher LSPR intensity. Besides, there is unrevealed absorption enhancement arisen at the region from 400 nm to 450 nm, which increases the light absorption ability of the thin film.

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## References

- [1] O'Regan B and Grätzel M, A low-cost, high-efficiency solar cell based on dye-sensitized colloidal TiO<sub>2</sub> films, *Nature* 353 (1991), 737-740.
- [2] Grätzel M, Photoelectrochemical cells, *Nature*, 414 (338) (2001), 338 – 344.
- [3] Martin A. Green, Thin-film solar cells: review of materials, technologies and commercial status, *J Mater Sci: Mater Electron*, 18 (2007), S15–S19.
- [4] S. Pillai n, M.A.Green, Plasmonics for photovoltaic applications, *Solar Energy Materials & Solar Cells*, 94 (2010), 1481–1486.
- [5] Yang Tian and Tetsu Tatsuma, Mechanisms and Applications of Plasmon-Induced Charge Separation at TiO<sub>2</sub> Films Loaded with Gold Nanoparticles, *J. AM. CHEM. SOC*, 127 (2005), 7632- 7637.
- [6] Harry a. Atwater and Albert Polman; Plasmonics for improved photovoltaic devices, *nature materials*, 9 (2010), 205 - 2013.
- [7] Wei Hao Lai, Yen Hsun Su, Lay Gaik Teoh , Min Hsiung Hon, Commercial and natural dyes as photosensitizers for a water-based dye-sensitized solar cell loaded with gold nanoparticles, *Journal of Photochemistry and Photobiology A: Chemistry*, 195 (2008), 307–313.
- [8] Stacey D. Standridge, George C. Schatz, and Joseph T. Hupp, Toward Plasmonic Solar Cells: Protection of Silver Nanoparticles via Atomic Layer Deposition of TiO, *Langmuir*, 25 (5) (2009), 2596-2600.
- [9] Changwoo Nahm, Hongsik Choi, Jongmin Kim, Dae-Ryong Jung, Chohui Kim, Joonhee Moon, Byungjoo Lee, and Byungwoo Park, The effects of 100 nm-diameter Au nanoparticles on dye-sensitized solar cells, *Appl. Phys. Lett.* 99 (2011), 253107.
- [10] Yen-Hsun Su, Yuan-Feng Ke, Shi-Liang Cai and Qian-Yu Yao, Surface plasmon resonance of layer-by-layer gold nanoparticles induced photoelectric current in environmentally-friendly plasmon-sensitized solar cell, *Light: Science & Applications*, 1 (2012), e14; doi:10.1038/lssa.2012.14 \_ 2012 CIOMP. All rights reserved 2047-7538/12
- [11] Vivian E. Ferry, Jeremy N. Munday, and Harry A. Atwater, Design Considerations for plasmonic Photovoltaics, *Adv. Mater*, 22 (2010), 4794–4808.
- [12] Tanujjal Bora, Htet H. Kyaw<sup>1</sup>, Soumik Sarkar, Samir K. Pal, and Joydeep Dutta, Highly efficient ZnO/Au Schottky barrier dye-sensitized solar cells: Role of gold nanoparticles on the charge-transfer process, *Beilstein J. Nanotechnol*, 2 (2011), 681–690.
- [13] Van Hieu Nguyen and Bich Ha Nguyen, Visible light responsive titania-based nanostructures for photocatalytic, photovoltaic and photoelectrochemical, applications ; *Adv. Nat. Sci.: Nanosci. Nanotechnol.* 3 (2012) 023001 (9pp) doi:10.1088/2043-6262/3/2/023001.
- [14] Dao Khac An, Pham Duy Long, Phan Anh Tuan, Vu Van Cat, Ngo Thi Hong Le, Nguyen Thi Mai Huong, Nguyen Duy Thien and LeVan Vu, On the plasmonic nano solar cells and several preliminary research results, invited talk (K-V1-12), The 6th International Workshop on Advanced Materials Science and Nanotechnology (IWAMSN2012) - October 30-November 02, 2012- Ha Long City, Vietnam.
- [15] Yu Zou, Zhi-An Wang, Xiao-Hua Lan and Ning-Kang Huang Anatase Phase TiO<sub>2</sub> Anode of a Dye-Sensitized Solar Cell Prepared by using RMFMS, *Journal of the Korean Physical Society*, Vol. 55, No. 6, December 2009, pp. 2650-2653.
- [16] Nguyen Duy Thien, Chu Dinh Kiem, and Nguyen Ngoc Long, Synthesis of Gold Nanorods from Metallic Gold by a Sonochemical Method, *e-J. Surf. Sci. Nanotech.* Vol. 9 (2011) 466-468.
- [17] Dufort C.C and Dragnea B, *Annu. Rev. Phys. Chem*, 61 (2010), 323-344.
- [18] K.R. Catchpole and A. Polman, Plasmonic solar cells, 16 (26) (2008), 21793 – 21800
- [19] Liz – Marzan L. M, Nano metals: Formation and color, *Materialstoday*, 2004, 28-31.
- [20] Jenó Gubicza, Janos L. Labar, Luu Manh Quynh, Nguyen Hoang Nam, Nguyen Hoang Luong. *Materials Chemistry and Physics*, 138 (2013), 449 – 453.